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#Jenny



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#Markus Jensen



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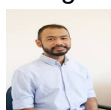
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Roll No. _____ Answer Sheet No. _____
Sig. of Candidate _____ Sig. of Invigilator _____

CHEMISTRY SSC-I SECTION - A (Marks 12)

Time allowed: 20 Minutes. Revised Syllabus
NOTE: Section A is compulsory. All parts of this section are to be answered on the question paper itself. It should be completed in the first 20 minutes and handed over to the Centre Superintendent. Detaching/governing is not allowed. Do not use lead pencil.

- Q.1 Circle the correct option i.e. A / B / C / D. Each part carries one mark.
- (1) Which of the following is a molecular ion?
A. CM^- B. CN^- C. N^+ D. O_2^+
 - (2) Which of the following isotopes is used to trace the flow of blood and detect possible constrictions or obstructions in the circulatory system?
A. Cobalt-60 B. Iodine-123 C. Carbon-14 D. Na-24
 - (3) Atomic Numbers were discovered by _____
A. Rutherford B. Moseley C. Dalton D. Bohr
 - (4) In which group of the Periodic table is d^5s^2 present on the basis of electronic configuration?
A. VI^A group B. VI^A group C. VI^A group D. VI^A group
 - (5) Due to high ionization energies, which elements do NOT lose electrons?
A. Metals B. Non-metals
C. Transition elements D. Radio active elements
 - (6) Which of the following is Sublime?
A. HCl (Acid) B. H_2SO_4 (Acid) C. Benzoic Acid D. Calcium Carbonate
 - (7) Melting point of Sodium Chloride is _____
A. 701°C B. 801°C C. 901°C D. 1001°C
 - (8) Vegetable oil is converted into vegetable ghee by passing H_2 gas through the oil. In this process which of the following elements is used as a catalyst?
A. Pt B. Fe C. Ni D. Cu
 - (9) What is the oxidation state of Hydrogen in case of Metallic Hydrides?
A. +1 B. +1 C. +2 D. -2
 - (10) Which of the following is used as Cathode in the Down's cell?
A. Iron B. Platinum C. Graphite D. Aluminium
 - (11) Which of the following is NOT of Magnesium?
A. MgO B. $Mg(OH)_2$ C. $MgCO_3$ D. $Mg(NO_3)_2$
 - (12) In Aqua Regia the ratio between HCl and HNO_3 is _____
A. 3 : 1 B. 2 : 1 C. 1 : 3 D. 1 : 2

For Examiner's use only:
Total Marks: 12
Marks Obtained: _____

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