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Quiz & Worksheet - How to Assign Oxidation Numbers to Elements in a Chemical Formula

https://study.com/academy/practice/quiz-worksheet-how-to-assign-oxidation-numbers-to-elements-in-a-chemical-formula.html

1. What are the oxidation numbers of the atoms in the compound CF_4 ?

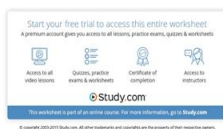
- a. +4, -4
- b. +4, -1
- c. +1, -4
- d. +1, -3

2. What is the oxidation number of each atom in KH ?

- a. +1, 1
- b. not enough information given.
- c. +1, -1
- d. -1, 1

3. What is an oxidation number?

- a. A number assigned to atoms in molecules to show the exact distribution of the electrons.
- b. A number assigned to atoms in molecules to show the exact distribution of the protons.
- c. A number assigned to atoms in molecules to show the general distribution of the electrons.
- d. A number assigned to atoms in molecules to show the general distribution of the protons.
- e. A number that tells you exactly which atom has how many electrons at any given time.



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