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Chapter 02 Testbank

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1. Anything that occupies space and has mass is called:
A. an electron
B. matter
C. energy
D. space
2. The electrons of an atom are:
A. proportional to the number of neutrons in an atom.
B. used to determine atomic number.
C. found in the nucleus.
D. not tied to particular orbital shells.
E. not tied to particular orbital shells.
3. All of the following pertain to the atom ^{14}C except:
A. has 6 electrons.
B. has 6 neutrons.
C. has 6 protons.
D. is an isotope of carbon.
E. mass number is 14.
4. The subatomic particles that surround the nucleus are the:
A. electrons.
B. protons.
C. neutrons.
D. protons and neutrons.
E. protons and electrons.
5. Cations are:
A. charged subatomic particles.
B. negatively charged electrons.
C. positively charged.
D. unstable of forming ionic bonds with anions.
E. atoms without protons.
6. Isotopes are atoms of the same element that differ in their:
A. atomic number.
B. proton number.
C. neutron number.
D. mass number.
E. chemical properties.
7. What is the maximum number of electrons in the second energy shell of an atom?
A. 4
B. 2
C. 18
D. 32
8. Two or more atoms bonded together are called a(n):
A. ion.
B. molecule.
C. element.
D. molecule.
E. molecule.
9. What would be the valence number of electrons in the sulfur atom ^{32}S ?
A. 6
B. 16
C. 18
D. 32
10. Polar molecules are composed of covalently bonded:
A. ionic atoms.
B. carbon atoms.
C. water.
D. atoms of different electronegativity.
E. atoms of identical electronegativity.

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