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ATOMIC STRUCTURE WORKSHEET

Complete the table without your periodic table.

Symbol	Atomic Number	Mass Number	Number of Protons	Number of Electrons	Number of Neutrons
Na			11		12
K		39		19	
			38		50
F				9	10
	20	40		20	
	50			50	69
I	53	127			
Mg			12		12
		108	47	47	
	1	1		1	
S		32	16		

Label the parts of the atom.

Which atomic particle determines the identity of the atom?

Atomic Structure
(Use your notes to answer the following)

1. _____ are the building block of all matter. Matter is anything that has _____ and takes up _____.

2. All atoms are made up of smaller subatomic particles called _____.

3. _____ Even though these particles are so small that we can't see them with our eyes, they still have masses called _____.

4. _____ Each particle has its own mass. For example protons and neutrons have an AMU of _____, but electrons are 2000 times smaller so their AMU is _____.

5. Every atom has an atomic number, which tells you the number _____ of _____ or _____ Atoms

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